



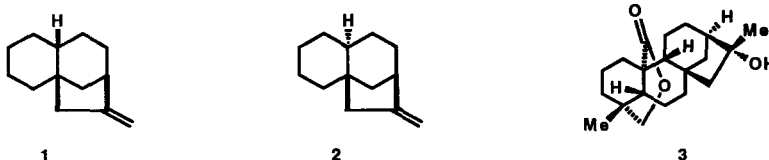
New Stereospecific Synthetic Routes to the Bicyclo[3.2.1]octane Subunit of the Kaurenoids and Gibberellins

E. J. Corey,* and Kun Liu

Department of Chemistry and Chemical Biology, Harvard University, Cambridge, Massachusetts 02138

Summary: Three new sequences are described for the synthesis of the bicyclo[3.2.1]octane core of diterpenoids such as the kaurenes (Schemes 1-3). © 1997 Elsevier Science Ltd.

The tricyclic system **1** is a prominent feature of many naturally occurring diterpenoids, including kaurenoids and gibberellins (*ent*-kaurenoids). The biological importance and structural complexity of these natural products have stimulated an enormous body of research on their synthesis.¹ In connection with recent studies directed at the total synthesis of the potent anti-HIV kaurenoid neotripterifordin (**3**),² we have developed new stereocontrolled processes for the construction of **1** and also the *trans*-6/6-fused diastereomeric unit **2**, which are described herein.



The first approach to the *cis*-6/6-fused subunit **1** is outlined in Scheme 1. Conjugate addition of the vinyl Gilman reagent to enone **4** produced stereospecifically the *cis*-fused β -vinyl ketone **5** in 95% yield.³ Reduction of ketone **5** by LiAlH₄ followed by hydroxy protection produced TBS ether **6** in 94% yield. Hydroboration of the vinyl group in **6** and oxidation gave aldehyde **7** in 92% yield. One-carbon homologation of aldehyde **7** to alkyne **8** was achieved in high yield using the Seyferth-Gilbert reagent.⁴ The corresponding xanthate prepared from alcohol **8** was treated with *n*-Bu₃SnH-AIBN in toluene at 110 °C to generate the cyclohexyl radical which underwent efficient cyclization to form **1** in 93% isolated yield.⁵

The success of the cyclization **8** → **1** prompted the study of a more direct route from **4** to **1** and led to the four-step synthesis which is shown in Scheme 2. Reaction of α,β -enone **4** and 1,2-propadienyltriphenylstannane⁶ using TiCl₄ as catalyst in CH₂Cl₂ (-78 °C for 1 h and -40 °C for 16 h) afforded the *cis*-fused acetylenic conjugate adduct **9** stereospecifically in 82% yield at 50% conversion along with recovered **4**. Although this reaction could not be driven to completion, it provided a ready source of the acetylenic ketone **9** since **4** and **9** are easily separated by silica chromatography. This ketone was efficiently converted to **1** by the sequence: (1) reduction with NaBH₄-CH₃OH; (2) xanthate formation and (3) radical cyclization with *n*-Bu₃SnH-AIBN in toluene at 110 °C.

The conversion of α,β -enone **4** to the *cis*-fused propargylic conjugate adduct **9** represents the first instance of propargylic addition to this or any other structurally related cyclic α,β -enone. The reaction did not occur at all with 1,2-propadienyltri-*n*-butylstannane (some 1,2-carbonyl addition was observed with this reagent), 1,2-

propadienyltriphenylplumbane or 1,2-propadienyltrimethylsilane. It is clear that this potentially important reaction requires further study and development.⁷

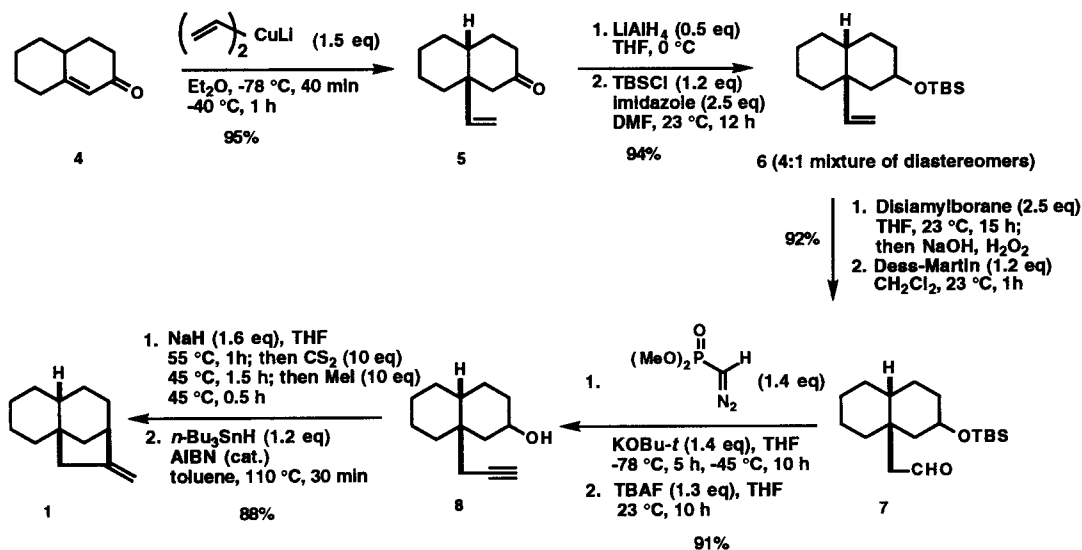
An efficient and stereocontrolled route to **2**, the tricyclic diterpene prototype containing a *trans* decalin subunit, was developed as outlined in Scheme 3. The photoaddition of allene to **1** occurred rapidly and cleanly (450 watt, medium-pressure Hanovia Hg lamp for use with an immersion well) to give an almost quantitative yield of the adduct **10**.⁸ Reduction of **10** with L-Selectride at -78 °C afforded stereospecifically the alcohol **11**.⁹ The orientation of the hydroxyl group in **11** was inverted by Mitsunobu displacement using benzoic acid and hydrolysis, and the resulting diastereomer of **11** was converted to the corresponding mesylate **12** (92% overall from **11**). Reaction of mesylate **12** with MeAlCl₂ at -78 °C in CH₂Cl₂ proceeded by rearrangement of the β-olefinic carbon to give the rearranged chloride **13** in essentially quantitative yield.¹⁰ Reductive dechlorination of **13** afforded **2**, the *trans*-6/6-fused diastereomer of **1** cleanly.

A different mode of rearrangement from **12** → **13** was observed with mesylate **14**, the diastereomer of mesylate **12**, as indicated in Scheme 4. Reaction of **14** with diethylaluminum bromide at -78 °C for 5 min provided in 91% yield the allylic bromide **15**,¹¹ the product of shifting the 6/4 fusion bond in a 1,2-carbon rearrangement. This interesting ring expansion reaction of an allene-α,β-enone photoadduct is a promising method for the synthesis of polycyclic terpenoids containing a fused pentalane subunit.

The observed rearrangements **12** → **13** and **14** → **15** can both be rationalized in terms of the assigned stereochemistry of **12** and **14** as concerted processes in which the migrating carbon moves backside to the carbon bearing the leaving group.

In summary the transformations outlined in Schemes 1-4 constitute useful new methodology for the synthesis of a variety of polycyclic structures, especially in the diterpenoid series.¹²

Scheme 1.



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9. The ^1H NMR data for alcohol **11** (carbinol CH-OH peak at 4.02 δ , dt, $J = 12.0, 6.0$ Hz) are consistent with a boat conformation of the six-membered ring in which the carbinol is a member. The boat arrangement is favored by the fusion with the four-membered ring. The stereochemistry assigned to **11** is also consistent with its formation by L-Selectride reduction of **10** and the ^1H NMR data for the epimeric alcohol: CHOH 3.92 δ (ddd, $J = 8.0, 6.0, 5.2$ Hz, 1 H).
10. The orientation of chlorine in **13** follows from the ^1H NMR spectrum; partial ^1H NMR data (500 MHz, CDCl_3): HCCl , 3.85 δ , s; C=CH_2 , 4.95 and 4.93 δ ; allylic CH 2.82 δ br s; allylic CH_2 2.48 δ (d, $J = 12$ Hz, 1 H) and 2.18 δ (dt, $J = 12.0, 1.9$ Hz, 1 H). The HCCl and bridgehead HC dihedral angle for **13** is estimated to be 86°, whereas that for the diastereomeric chloride is expected to be 34° (from measurements with HGS models).
11. Partial ^1H NMR data for allylic bromide **15** (500 MHz in CDCl_3): one olefinic proton at 5.75 δ (s, 1 H); CH_2Br at 4.01 δ (s, 2 H); HC-C=C at 2.45 δ (m, 1 H); $\text{H}_2\text{CC=C}$ at 2.21 δ (d, $J = 12.0$ Hz, 1 H) and 2.06 δ (d, $J = 12.0$ Hz, 1 H).
12. This research was supported by grants from the National Institutes of Health and the National Science Foundation.

(Received in USA 31 July 1997; revised 22 August 1997; accepted 25 August 1997)